

# Practical Chemistry

HazMat ID for HHW Personnel  
Josh Prouty

1

## What is Hazardous?

Hazardous materials generally have one or more properties that make them hazardous.

- Flammable or Combustible
- Corrosives (Acidic or Alkaline)
- Reactive (Fumes or Heat when mixed)
- Toxic (Poisonous to humans or animal life)

Not all labels are created equal and product could still meet definition of hazardous.

- Non-toxic- Could still be flammable i.e. hand sanitizer or corrosive i.e. vinegar cleaner.
- Biodegradable- Likely to be corrosive, many lye-based drain cleaners are biodegradable.
- Organic- Could be toxic i.e. copper based fungicide used in organic farming.
- When in doubt look at SDS for active ingredients.

2

2

# Packaging Identification

What's in a bottle

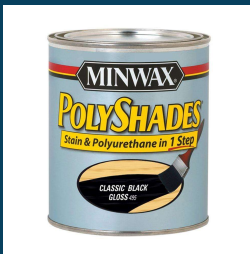
3

3

## Packaging Types

- Packaging can reveal a lot about the physical properties even without a label.
- Oil-Based Paints will not likely be found in plastic containers due to solvents eating the container. Key Words: Combustible, Oil-based, Solvent, Alkyd)

Water-based paint can be in either plastic or metal containers.  
Key words: Latex, Acrylic, Water-based, or water clean up  
When in doubt assume its hazardous i.e. oil-based.



4

4

## Corrosives

- Strong Corrosives are sometimes in color coded glass or plastic bottles.
- Typically they are in distinctive containers that are similar regardless of the brand.
- Cleaners of all types are typically either a solvent (flammable or a corrosive

Acids-Mostly Red Cap



Alkaline- Blue Cap



5

## Cleaners

- Rust, lime, and calcium removers are typically a Mild Acid because minerals naturally dissolve in acids.
- All Purpose, Bathroom, and Kitchen Cleaners are typically a Mild Alkaline solutions which foam/bubble when shaken.

Mild Acid



Mild Alkaline



6

6

## Oxidizers

- Some oxidizers can also be identified by their package.



Liquid Pool Chlorine often comes in this distinctive yellow container with the active ingredient being Sodium Hypochlorite >10%.



Many solid oxidizers (pool shock tabs) come in these distinctive containers with ratchet or safety top lids. Non-Chlorine Bleach like Oxiclean, also come in a distinctive laundry detergent style cleaner.

7

7

## Poisons

- Some poisons can be identified by their distinctive packaging or cap.



Hydramethalnon (Amdro) is unique in using this shaker top plastic container.



Chevron/Ortho pioneered the use of this yellow cap in a brown bottle for their pesticide line that is now used by many others.

The brown glass moonshine style bottle is also a good indicator that this is likely an old pesticide

8

8

## Petroleum, Oil, and Lubricants (POL)

Many automotive products (POL) can be identified by their distinctive packaging or appearance



Antifreeze will often be in these type of 1-gallon jugs with a safety cap. Fluid will also be dyed a bright color like orange, green, red, or yellow.

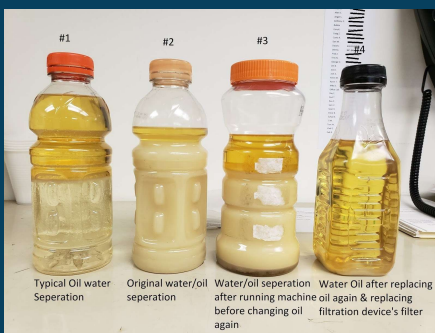
Oil products will often be in these distinctive 1 or 5 qt or even 1pt containers.

9

9

## Petroleum Oil and Lubricants (POL) Continued

Many automotive products (POL) can be identified by their distinctive packaging or appearance



Water contaminated oil will often have a mayonnaise type appearance due to forming an emulsion.









Dirty antifreeze can vary in color and consistency depending on contamination

10

10






## Battery by shape and type

<p><b>Ni-MH (Nickel Metal Hydride)</b></p>	<p>Nickel Metal Hydride performs well in high drain devices and can be recharged up to 1,000 times. It's cadmium-free, but can be pricey.</p>	
	<p>Ni-MH batteries can typically be found in cordless power tools, digital cameras, two-way radios and cordless phones.</p>	
<p><b>Ni-Cd (Nickel Cadmium)</b></p>	<p>Nickel Cadmium is the least expensive of secondary/rechargeable batteries, and can be recharged up to 1,000 times. Cadmium can be toxic to the environment and should be recycled.</p>	
	<p>Ni-Cd can typically be found in cordless power tools, digital cameras, two-way radios and cordless phones.</p>	
<p><b>SSLA/Pb (Small Sealed Lead Acid)</b></p>	<p>Small Sealed Lead Acid (SSLA/Pb) batteries are simple to manufacture and have one of the lowest discharge rates of any rechargeable battery.</p>	
	<p>SSLA/Pb can typically be found in emergency devices, emergency exit signs, security systems, mobility scooters and UPS back-ups.</p>	

11

11

## Batteries Continued

<p><b>Li-Ion (Lithium Ion)</b></p>	<p>Lithium Ion has excellent performance in high drain devices, though it can be considered expensive.</p>		
	<p>Li-Ion can typically be found in cell phones, laptops, two-way radios, and cordless power tools.</p>		
<p><b>Primary/Single Use</b></p>	<p>Primary/Single-Use batteries cannot be recharged and must be disposed of once they wear out. Alkaline, lithium, silver oxide, zinc-air, zinc-carbon, and zinc-chloride, commonly known as AA, AAA, 9V, D-cell, and button cell are all types of primary batteries.</p>		
	<p>Primary/Single-Use batteries are most commonly used in low-drain devices, such as flashlights, clocks, watches, remote controls, and smoke detectors.</p>		
	<p><b>NiMH Cells from hybrid cars</b></p>		
			
			<p><b>Wet Cell NiCad- These are specialty batteries that are used in the aerospace industry and require special handling with only a few recyclers accept them.</b></p>

12

12

# Label Identification

What's in a name

13

13

## Chemistry Systematic Naming Functional Groups

### Systematic Naming

- Organic chemistry uses a naming system of prefixes and suffixes that can help identify the makeup of a chemical.
- Prefixes are typically number-based. Mono/Un=1, Di/Bi=2, Tri/Pro/Tert=3, Quad/Bu/Tetra=4, etc.
- Suffixes typically identify the functional group. –ol=Alcohols, -ene=Alkene
- Inorganic Chemistry suffixes indicate functional group or oxidation state.
- -ide=binary compounds NaCl, KI, HCl, etc.
- -ite=1-2 oxygen atoms on ion NaHClO hypochlorite, NaHClO<sub>2</sub> chlorite
- -ate=3-4 oxygen atoms MgClO<sub>3</sub> chlorate, K<sub>2</sub>CrO<sub>4</sub> chromate, NaClO<sub>4</sub>

### Functional Groups

- Functional groups can be considered classes of chemicals that share common properties useful to haz mat identification.
- A knowledge of the most common functional groups and their suffixes can allow you to identify properties of an unknown chemical just from the name.
- Ex. Isopropanol= ISO=Isomer or non standard form, Propan= 3 carbon chain, -ol=alcohol. A 3 carbon alcohol in a non standard formation.

14

14

# Functional groups in organic chemistry



Functional groups are the characteristic groups in organic molecules that give them their reactivity. In the formulae below, R represents the rest of the molecule and X represents any halogen atom.

● Hydrocarbons ● Halogen-containing groups ● Oxygen-containing groups ● Nitrogen-containing groups ● Sulfur-containing groups ● Phosphorus-containing groups



**Alkane**  
Naming: -ane  
e.g. ethane



**Alkene**  
Naming: -ene  
e.g. ethene



**Alkyne**  
Naming: -yne  
e.g. ethyne



**Arene**  
Naming: -yl benzene  
e.g. ethyl benzene



**Haloalkane**  
Naming: halo-  
e.g. chloroethane



**Alcohol**  
Naming: -ol  
e.g. ethanol



**Aldehyde**  
Naming: -al  
e.g. ethanal



**Ketone**  
Naming: -one  
e.g. propanone



**Carboxylic acid**  
Naming: -oic acid  
e.g. ethanoic acid



**Acid anhydride**  
Naming: -oic anhydride  
e.g. ethanoic anhydride



**Acyl halide**  
Naming: -oyl halide  
e.g. ethanoyl chloride



**Ester**  
Naming: -yl -oate  
e.g. ethyl ethanoate



**Ether**  
Naming: -oxy -ane  
e.g. methoxyethane



**Epoxide**  
Naming: -ene oxide  
e.g. ethene oxide



**Amine**  
Naming: -amine  
e.g. ethanamine



**Amide**  
Naming: -amide  
e.g. ethanamide



**Nitrate**  
Naming: -yl nitrate  
e.g. ethyl nitrate



**Nitrite**  
Naming: -yl nitrite  
e.g. ethyl nitrite



**Nitrile**  
Naming: -nitrile  
e.g. ethanenitrile



**Nitro**  
Naming: nitro-  
e.g. nitromethane



**Nitroso**  
Naming: nitroso-  
e.g. nitrosoethane



**Imine**  
Naming: -imine  
e.g. ethanimine



**Imide**  
Naming: -imide  
e.g. succinimide



**Azide**  
Naming: -yl azide  
e.g. phenylazide



**Cyanate**  
Naming: -yl cyanate  
e.g. methyl cyanate



**Isocyanate**  
Naming: -yl isocyanate  
e.g. methyl isocyanate



**Azo compound**  
Naming: -azo-  
e.g. azoethane



**Thiol**  
Naming: -thiol  
e.g. methanethiol



**Sulfide**  
Naming: sulfide  
e.g. dimethyl sulfide



**Disulfide**  
Naming: disulfide  
e.g. dimethyl disulfide



**Sulfoxide**  
Naming: sulfoxide  
e.g. dimethyl sulfoxide



**Sulfone**  
Naming: sulfone  
e.g. dimethyl sulfone



**Sulfinic acid**  
Naming: -sulfinic acid  
e.g. benzenesulfinic acid



**Sulfonic acid**  
Naming: -sulfonic acid  
e.g. benzenesulfonic acid



**Sulfonate ester**  
Naming: -yl sulfonate  
e.g. methylmethanesulfonate



**Thiocyanate**  
Naming: thiocyanate  
e.g. ethyl thiocyanate



**Isothiocyanate**  
Naming: isothiocyanate  
e.g. ethyl isothiocyanate



**Thial**  
Naming: -thial  
e.g. ethanethial



**Thio ketone**  
Naming: -thione  
e.g. propanethione

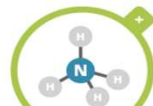


**Phosphine**  
Naming: phosphane  
e.g. methylphosphane

15

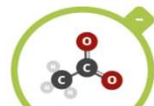
## POLYATOMIC IONS: NAMES, FORMULAE & CHARGES

A polyatomic ion is a charged species consisting of two or more atoms covalently bonded together. Here's a guide to some of the most common examples!



**AMMONIUM**

Formula:  $\text{NH}_4^+$



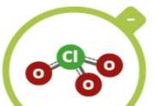
**ACETATE**

Formula:  $\text{C}_2\text{H}_3\text{O}_2^-$



**CARBONATE**

Formula:  $\text{CO}_3^{2-}$



**CHLORATE**

Formula:  $\text{ClO}_3^-$



**CHLORITE**

Formula:  $\text{ClO}_2^-$



**CHROMATE**

Formula:  $\text{CrO}_4^{2-}$



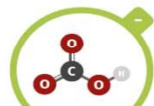
**CYANIDE**

Formula:  $\text{CN}^-$



**DICHROMATE**

Formula:  $\text{Cr}_2\text{O}_7^{2-}$



**HYDROGEN CARBONATE**

Formula:  $\text{HCO}_3^-$



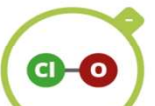
**HYDROGEN SULFATE**

Formula:  $\text{HSO}_4^-$



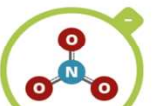
**HYDROXIDE**

Formula:  $\text{OH}^-$



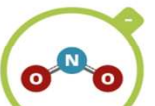
**HYPOCHLORITE**

Formula:  $\text{ClO}^-$



**NITRATE**

Formula:  $\text{NO}_3^-$



**NITRITE**

Formula:  $\text{NO}_2^-$



**PERCHLORATE**

Formula:  $\text{ClO}_4^-$



**PERMANGANATE**

Formula:  $\text{MnO}_4^-$



**PEROXIDE**

Formula:  $\text{O}_2^{2-}$



**PHOSPHATE**

Formula:  $\text{PO}_4^{3-}$



**SULFATE**

Formula:  $\text{SO}_4^{2-}$



**SULFITE**

Formula:  $\text{SO}_3^{2-}$



**THIOSULFATE**

Formula:  $\text{S}_2\text{O}_3^{2-}$

16

## Properties of Functional Groups

### Flammable Functional Groups

#### Common Strongly Flammable Functional Groups

- Alkanes –ane
- Alkene –ene
- Alkyne –yne
- Alkyd- a type of polyester (**a**lcohol + organic **a**cid)
- Alcohol –ol, glycol/diol (2 alcohol groups), glycerin/glycerol (3 alcohol groups)
- Ethers –ether i.e. Dimethyl ether
- Ketone –one
- Haloalkane- ie. Chloromethane, Chloroform

### Corrosive Functional Groups

#### Common functional groups that are acidic or basic

- Carboxylic acid –oic acid
- Amine –amine
- Amide –amide
- Aldehyde- al (ex. propanal)

17

17

## Properties of Functional Groups

### Reactive Functional Groups

#### Common Reactive Functional Groups

- Aldehyde –al or aldehyde
  - Ether –ether
  - Cyclic Ether –Tetrahydrofuran, or oxolane
- Nitrogen containing compounds are often reactive
- Nitrate –nitrate NO<sub>3</sub>
  - Nitrite –nitrite NO<sub>2</sub>
  - Isocyanate –isocyanate Potassium Isocyanate(KNCO)
  - Fulminate Potassium Fulminate(KCNO)
  - Azide –azide Sodium Azide (NaN<sub>3</sub>)
  - Amine –amine NH<sub>3</sub>
  - Amide -amide

### Toxic Functional Groups

#### Common functional groups that are toxic and could release fumes.

- Cyanides
- Cyanates Potassium Cyanate (KOCN)
- Dioxins
- Arsenate
- Haloalkane - Chloroform

18

18

## Common Pesticides and Naming

**Pesticides is the generic term for all substances that are meant to be toxic when applied to a pest species, weed, or other organism.**

### Common Pesticides

- DDT (banned)
- Deet
- Permethrin
- Carbaryl (currently being relabeled)
- Lindane (banned)
- Chlordane (banned)
- Nicotine
- Bifenthrin
- Pyrethrins
- Chlorpyrifos (banned 2021)
- Malathion
- Zinc Phosphide
- Arsenic baits

### Common Herbicides

- 2-4-D
- 2-4-5-T (banned)
- Glyphosate
- Triclopyr
- Atrazine
- Picloram
- Paraquat

19

19

## Common Pesticides and Naming

**Pesticides is the generic term for all substances that are meant to be toxic when applied to a pest species, weed, or other organism.**

### Common Algaecide

- Simazine
- Benzalkonium Chloride
- Copper Sulfate

### Common Fungicide

- Chlorothalonil
- Captan
- -azole (fluconazole)
- Liquid Copper compounds (Copper Sulfate)

20

20

## Common Corrosives and Naming

**Corrosives are anything that has a pH less than 5 or more than 8. These include many cleaning chemicals. Beware Biodegradable doesn't mean non-hazardous.**

### Common Alkaline

- Most Drain Cleaners (Draino, Red Devil)
- **Beware of acid drain cleaners (Hair Clog Remover)**
- Lye (Sodium Hydroxide)
- Potash (Potassium Hydroxide)
- Simple Green
- Most All-purpose cleaners
- Bleach <10% (Sodium hypochlorite)

### Common Acid

- Hydrochloric Acid (HCL)- Muriatic Acid
- Phosphoric Acid Cleaner
- Sulfuric acid drain cleaner
- Acidic Tile Cleaner
- CLR (Calcium Lime Rust) Cleaner
- Naval Jelly/ Rust Remover Jelly

21

21

## Common Oxidizers and Naming

**Oxidizers generally fit into a couple of common categories and must be segregated by chemical (i.e. Chlorine vs Bromine vs Peroxide). Pool Cleaners, Gardening Chemicals, and Industrial Chemicals. Notice that oxygen atoms are a common theme in name/formula of most oxidizers.**

### Common Pool Oxidizers

- Sodium hypochlorite >10%
- trichloro-s-triazinetrione tablets (Trichlor)
- Potassium Monopersulfate, or Potassium Peroxymonosulfate (Non Chlorine Shock)
- 1-Bromo-3-Chloro-5,5-dimethylhydantoin (Bromine Tabs)

### Common Gardening

- Potassium Nitrate Fertilizer
- Stump Out (KNO<sub>3</sub>)

### Industrial Chemical/Cleaner

- Nitric Acid –HNO<sub>3</sub>
- Hydrogen Peroxide H<sub>2</sub>O<sub>2</sub> >3%
- Sodium Percarbonate (Oxyclean)
- Peroxides
- Silver Nitrate
- Methyl Ethyl Ketone Peroxide (MEKP)

22

22

**ALACHUA COUNTY**  
1824-2024 | BICENTENNIAL

## Questions?

**Joshua Prouty, CPM**  
Hazardous Waste Coordinator  
Solid Waste and Resource Recovery  
5125 NE 63rd Ave • Gainesville • FL • 32609  
352-334-0440 (office)

DISCOVER **200** | **ALACHUA COUNTY**  
1824-2024 | BICENTENNIAL

23